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# Co dependence of Curie temperature in amorphous Fe–Co–Zr–B–Nb alloys with high glass-forming ability

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#### Abstract

Effects of Co substitution for Fe on the Curie temperature ( $T_c$ ), glassforming ability (GFA) and thermal stability of amorphous Fe<sub>61-x</sub>Co<sub>x</sub>Zr<sub>5</sub>B<sub>30</sub>Nb<sub>4</sub> (FCZBN) alloys were studied for Co content ranging from 0 to 15 at. %. The  $T_c$  shows a sinusoid-like behaviour with increasing Co content, revealing two maxima at 3 and 12.5 at. % Co and a minimum at 7.5 at. % Co. Co content dependences of glass transition ( $T_g$ ), crystallization ( $T_x$ ) and reduced glass transition temperatures ( $T_{rg}$ ) of the amorphous alloys are almost completely opposite to that of the  $T_c$ . The  $T_c$  decreases with increasing  $T_g$  and  $T_{rg}$ , but increases with increasing Co content. The Co content dependence of the  $T_c$  is suggested to relate to both Co content and high GFA of the amorphous alloys.

### 1. Introduction

Effects of Co substitution for Fe on thermal stability, glass-forming ability (GFA) and magnetic properties of Fe-based amorphous alloys have been studied widely, owing to excellent magnetic properties of the amorphous alloys. It was found that the effect of substitution of Co for Fe on the Curie temperature  $T_c$  was different for various Fe-based amorphous alloys. For some Fe-based amorphous alloys, this substitution makes the  $T_c$  either increase or decrease monotonically with increasing Co content [1–5]. For other Fe-based amorphous alloys, the  $T_c$  firstly increases up to a maximum with increasing Co content and then decreases [6–9]. The effects of this substitution on the  $T_c$  are usually attributed to changes of atomic moments of Fe and Co and/or exchange interactions of Fe–Fe, Fe–Co and Co–Co atomic couples [1–9].

It is well known that exchange interaction is related to species and interatomic distance of the nearest-neighbour magnetic atoms in amorphous alloys, based on the approximation of nearest-neighbour atoms. Since the density of random-packed atoms determines the interatomic distance, it should affect the exchange interaction of magnetic atoms in a magnetic amorphous alloy. For conventional Fe–Co based amorphous alloys, variation in exchange interaction induced by change of interatomic distance may be much smaller than that induced by change of magnetic atomic couples when the Fe atom is replaced by the Co atom, and hence the effect of the interatomic distance on the  $T_c$  is not obvious. However, amorphous alloys with higher GFA, glass transition ( $T_g$ ) and crystallization ( $T_x$ ) temperatures, have much denser random-packed microstructures than the conventional amorphous alloys [10–13], leading to larger variation in the interatomic distance. This larger variation will cause a considerable influence on the exchange interaction, so that the effect of interatomic distance on the  $T_c$ becomes obvious.

Recently, it was reported that the GFA of Fe–Zr–B-based alloy could be greatly improved by the addition of some transition metals, such as Co, Ni, Ta, Nb, W, Mo, etc [14–17], so that some bulk Fe–Zr–B-based amorphous alloys with high GFA and thermostability were prepared at low cooling rate. However, their magnetic properties were different [14, 15]. Some of them are ferromagnetic at room temperature, while others are paramagnetic, which seems to be related to their GFA [15–17]. Thus, it is necessary to study correlations between GFA,  $T_g$  and  $T_c$ . However, there have been no detailed studies of the correlations so far.

In this paper, effects of Co substitution for Fe on GFA, thermal stability and magnetic properties were studied for the amorphous Fe–Zr–B–Nb alloys, and correlations between thermostability, GFA and magnetic properties were investigated in detail.

#### 2. Experimental procedures

Multicomponent alloys with nominal atomic ratios of  $Fe_{61-x}Co_xZr_5B_{30}Nb_4$  (FCZBN) (x = 0-15) were prepared by arc-melting Fe, Co, Zr, B and Nb with purity ranging from 99.5 to 99.98% in a titanium-gettered argon atmosphere. Amorphous FCZBN ribbons (about 1.5 mm in width and 25  $\mu$ m in thickness) were prepared by single roller melt-spinning technique in an argon atmosphere. The microstructure of the ribbons was characterized by x-ray diffractometry (XRD) with Cu K<sub> $\alpha$ </sub> radiation and transmission electron microscopy (TEM) to be fully amorphous alloys. The onset melting ( $T_m$ ) and liquidus ( $T_1$ ) temperatures of the FCZBN alloys were determined by a differential temperature analyser at a heating rate of 20 °C min<sup>-1</sup>. Measurements of  $T_g$ ,  $T_x$  and  $T_c$  were performed by differential scanning calorimetry (DSC) at a heating rate of 20 °C min<sup>-1</sup> in a flowing argon atmosphere. Apparent activation energy was measured by means of DSC at heating rates of 2.5, 5, 10, 20 and 40 °C min<sup>-1</sup>. The  $T_c$  was also confirmed by measuring the change in magnetization with temperature using a vibrating sample magnetometer (VSM) at a heating rate of 0.1 °C min<sup>-1</sup> in an external field of 500 Oe. Magnetization of the samples was measured by the VSM.

#### 3. Results and discussion

Since transition from a ferromagnetic to paramagnetic phase is a second-order phase transformation, an inflection appears at the  $T_c$  in each DSC curve of the amorphous FCZBN alloys, as shown in figure 1. The  $T_c$  of the amorphous FCZBN alloy with 7.5 at. % Co was also measured by VSM to be 241 °C, which is 11 °C higher than the  $T_c$  measured by DSC. However, when the effect of the external magnetic field on the measured value of the increase in  $T_c$  is taken into account, the  $T_c$  values measured by DSC and VSM are in agreement. As shown in figure 2(a), the  $T_c$  shows a sinusoid-like changing behaviour with increasing Co content. It increases again until the Co content reaches 12.5 at. % and finally decreases with increasing Co content up to 15 at. %. This tendency is very different from Co-content dependences of the  $T_c$  observed in conventional amorphous Fe–Co–Zr–B–Nb alloys with low B content [1, 5] and other Fe–Co-based amorphous alloys [6–9, 18]. Figure 2(b) reveals



Figure 1. DSC curves for the measurement of the Curie temperature of the as-spun amorphous FCZBN alloys with various Co contents.

Co-content dependence of magnetizations of the amorphous FCZBN alloys measured at room temperature ( $M_{\rm RT}$ ), which is similar to that of the  $T_c$ , implying that the Co-content dependence of the  $T_c$  measured by DSC is correct.  $T_g$ ,  $T_x$  and reduced glass transition temperature  $T_{\rm rg}$  ( $T_{\rm rg} = T_g/T_1$ ) versus Co content plots are shown in figure 2(c). It is well known that  $T_{\rm rg}$  is usually used to characterize GFA [19–21] of amorphous alloys, while  $T_g$  and  $T_x$  are used to characterize their thermal stability. Figure 2(c) indicates that the Co-content dependence of thermal stability is similar to that of GFA in the amorphous FCZBN alloys.

It is known that a very small amount of crystallite in an amorphous alloy may significantly affect the physical properties of the amorphous alloy, such as thermal stability and magnetic properties, etc. Although no crystallite is detected by XRD and TEM in the amorphous FCZBN ribbons, it is still necessary to study the effect of volume fraction transformed on the Curie temperature to confirm that the Co-content dependence of the  $T_c$  is due to instinctive properties of the amorphous FCZBN alloys or due to the effect of a very small amount of crystallite that was not detected. An amorphous FCZBN alloy with 5 at. % Co was run repeatedly in DSC in the following way: heating at 20 °C min<sup>-1</sup> from room temperature to 640 °C, then isothermally annealing for 5 min at this temperature and finally uncontrolled cooling to room temperature. Since the temperature 640 °C is located in a supercooled liquid region  $\Delta T_x$  ( $\Delta T_x = T_x - T_g$ ), the amorphous FCZBN alloy crystallizes very slowly and its volume fraction transformed increases gradually and slowly after each run. Figures 3(a)–(e) show typical DSC curves of amorphous FCZBN alloys recorded at the first, second, fourth, sixth and eighth runs respectively, indicating that after all runs the Curie temperature of the amorphous FCZBN alloy changed less than 1 °C, except after the first run when it increased 7 °C, which is attributed to relaxation and



**Figure 2.** Variations of  $T_c$  (a), magnetization (b) and  $T_g$ ,  $T_x$  and  $T_{rg}$  (c) of the amorphous FCZBN alloys as a function of Co content.  $\blacktriangle T_g$ ;  $\spadesuit T_x$ ;  $\blacksquare T_{rg}$ .

increase of chemical short-range order in the amorphous FCZBN alloy. However, intensity of the inflection at Curie point decreased slowly with increasing run times. Figures 4(a)–(d)show XRD profiles of the as-spun FCZBN ribbon and the ribbon run after the third, fifth and seventh times respectively. Obviously, the amorphous FCZBN alloy crystallizes gradually in a eutectic mode to  $\alpha$ -Fe (Co) and a Fe–Co–Zr–Nb–B phase, the structure of which is not yet identified. Its volume fraction transformed increases with increasing run times, whereas the volume fraction of the amorphous FCZBN alloy decreases and becomes very small after the seventh time. On comparison of figures 3 and 4, we infer that the intensity of the inflection decreases on decreasing the volume fraction of the amorphous alloy (or increasing the volume fraction transformed), and becomes very weak when the volume fraction of the amorphous alloy is very small. However, the Curie temperature changes very little. The above results indicate that the Curie temperature of the amorphous FCZBN alloys is not affected by the volume fraction transformed. The Co-content dependence of the  $T_c$ , shown in figure 2(a), is a natural property of the amorphous FCZBN alloy. The same conclusion was also obtained in the amorphous FCZBN alloy with 3, 7.5 and 12.5 at. % Co. In addition, it was also found that  $T_g$  and  $T_x$  do not change with the volume fraction transformed for the amorphous FCZBN ribbons.

It is well known that the Curie temperature depends on the atomic moment and exchange interaction of magnetic atoms. However, it was reported that the average atomic moments



Figure 3. DSC curves of the amorphous FCZBN alloy with 5 at. % Co recorded at the first (a), second (b), fourth (c), sixth (d) and eighth (e) runs.



**Figure 4.** XRD profiles of the amorphous FCZBN alloy with 5 at. % Co (a) and the amorphous alloy run after three (b), five (c) and seven times (d).  $\bullet \alpha$ -Fe (Co);  $\blacksquare$  Fe–Co–Zr–B–Nb.

of Fe and Co atoms change little in Co content; less than 20 at. % for Fe-rich Fe–Co-based amorphous alloys [1–4]. Therefore, the variation of  $T_c$  is dominated by exchange interaction in the amorphous FCZBN alloys.

Previous studies have reported that the  $T_c$  increases monotonically with increasing Co content in Fe–Co–Zr–B–Cu and Fe–Co–Zr–B–Nb–Cu amorphous alloys with low B content, which has been attributed to the fact that exchange interactions of Co–Co and Fe–Co atomic couples are stronger than those of the Fe–Fe atomic couple [1, 5]. However, this mechanism



**Figure 5.** Plots of  $T_c$  as a function of  $T_g$  (a),  $T_{rg}$  (b) and Co content (c) for the amorphous FCZBN alloys. The number beside each symbol in (a) and (b) represents Co content of the amorphous FCZBN alloy corresponding to the symbol, and in (c)  $\blacksquare$ , represents the amorphous FCZBN alloys with close  $T_{rg}$ .

hardly explains the change of  $T_c$  observed in the present experiment. Therefore, the  $T_c$  should be affected by additional factors besides exchange interactions of Co–Co, Fe–Co and Fe–Fe atomic couples in the amorphous FCZBN alloys.

Comparison of figures 2(a) and (c) shows that the Co-content dependence of the  $T_c$  is almost completely opposite to that of  $T_g$ ,  $T_x$  and  $T_{rg}$ , which implies that some correlation may exist between the  $T_c$  and  $T_g$ ,  $T_x$  and  $T_{rg}$  in the amorphous FCZBN alloys. Figure 5(a) shows a  $T_c$  versus  $T_g$  plot of the amorphous FCZBN alloys, the number beside each symbol represents Co content of the amorphous FCZBN alloy corresponding to the symbol. It indicates that the  $T_c$  decreases with increasing  $T_g$ . For amorphous alloys with close values of  $T_g$ , such as the amorphous alloys with Co contents of 2, 3, 4 and 10 at. %,  $T_c$ increases with increasing Co content. Figure 5(b) shows a plot of  $T_c$  as a function of  $T_{rg}$ , revealing that the  $T_c$  decreases with increasing  $T_{rg}$  in each of the four content ranges 0–3, 3-7.5, 7.5-12.5 and 12.5-15 at. % Co. But the decreasing rate of the  $T_c$  is different for different Co-content ranges. The rate is higher for Co contents ranging from 0 to 3 at. % and 7.5 to 12.5 at. %, for which  $T_{rg}$  increases with decreasing Co content, compared with Co contents of 3 to 7.5 at. % and 12.5 to 15 at. %, for which  $T_{rg}$  increases with increasing Co content. For the amorphous FCZBN alloys with close values or the same  $T_{rg}$ , such as the amorphous alloys with Co contents of 0, 2, 4, 10 and 15 at. %, as shown in figure 5(b), the  $T_c$  (indicated by  $\blacksquare$  in figure 5(c)) increases with increasing Co content monotonically, similar to previously reported results [1, 5]. By fitting the points marked by the solid squares, a  $T_c$  versus Co-content curve, shown by a broken line in figure 5(c), was obtained. It gives a tendency  $T_c$  with Co content for amorphous alloys having  $T_{rg}$  of about 0.58. It suggests that the  $T_c$  increases monotonically with increasing Co content for amorphous FCZBN alloys with the same or close values  $T_{rg}$ . However, for amorphous FCZBN alloys with  $T_{\rm rg}$  less than 0.58, such as the amorphous FCZBN alloys with Co content of 3 or 12.5 at. %, the  $T_c$  (denoted by  $\bullet$  in figure 5(c)) is always higher than the Curie temperature  $(T'_c)$  evaluated from  $T_c$  versus Co-content curve at its corresponding Co content, such as 3 or 12.5 at. % Co. For the amorphous FCZBN alloys with  $T_{rg}$  much larger than 0.58, such as the amorphous FCZBN alloy with Co content of 5 or 7.5 at. %,  $T_c$  (denoted by  $\blacktriangle$  in figure 5(c)) is far less than  $T_{\rm c}'$  evaluated from  $T_{\rm c}$  versus Co-content curve at the same Co content, such as 5 or 7.5 at. % Co. These results indicate that the  $T_c$  decreases with increasing  $T_{rg}$  for amorphous FCZBN alloys under the same Co content. It was also found from figure 5(c) that some amorphous FCZBN alloys with higher Co content and  $T_{rg}$  have less  $T_{c}$  than those with lower Co content and  $T_{rg}$ , while some amorphous FCZBN alloys with higher Co content and  $T_{rg}$  have a larger  $T_{c}$ compared with those having lower Co content and  $T_{rg}$ . For example, the  $T_c$  of the amorphous FCZBN alloy with 7.5 at. % Co and  $T_{rg}$  of 0.608 is less than that of the amorphous FCZBN alloys with Co content less than 5 at. % and  $T_{\rm rg}$  less than 0.58, while  $T_{\rm c}$  of the amorphous FCZBN alloy with 10 at. % Co and  $T_{rg}$  of 0.58 is larger than that of the amorphous FCZBN alloy with 3 at. % Co and  $T_{rg}$  of 0.57, as shown in figure 5(c). In the former,  $T_{rg}$  dominates  $T_c$ , implying that  $T_{rg}$  has more effect on the  $T_c$  than Co content. While in the latter, Co content dominates  $T_c$ , implying that Co content has more influence on  $T_c$  than  $T_{rg}$ . On the basis of the analysis mentioned above, it is concluded that the  $T_c$  of the amorphous FCZBN alloys is determined by Co content and  $T_{rg}$ , and that the  $T_c$  increases with increasing Co content but decreases with increasing  $T_{rg}$ . Co-content dependence of the  $T_c$  is determined by a competition of effects of Co content and  $T_{rg}$  on it.

It was reported that  $T_g$ ,  $T_x$  and GFA are related to the atomic interaction and the dense random-packed structure of amorphous alloys [10–13]. Many experimental results have demonstrated that an amorphous alloy with high  $T_g$ ,  $T_x$  and GFA has highly dense randompacked structure or high density [10–13]. Thus, it can be deduced from figure 2(c) that the amorphous FCZBN alloy with high  $T_{rg}$  should have a highly dense random-packed structure and the amorphous FCZBN alloy with 7.5 at. % Co should have the largest density. This can be attributed to the negative heats of mixing of Co–Zr and Co–Nb, which are much larger than those of Fe–Zr and Fe–Nb and differences in heat of mixing between Co–Zr(Nb) and Fe–Zr(Nb) change with atomic ratios of Fe and Co to Zr and Nb and reach maxima at atomic ratios of 1 : 1 to 1.5 : 1 [22]. For amorphous FCZBN alloys without Co, the atomic ratio of Fe to the number of B and Fe is about 2 : 3, and so, on average six Zr and Nb atoms should become nearestneighbour atoms of Fe. Co usually occupies the position of Fe in Fe–Zr (or Nb) clusters to form Co–Zr(Nb) clusters upon the substitution of Co for Fe, and the atomic ratio of Co to Zr and Nb is 1.25 : 1 in the amorphous FCZBN alloys with 7.5 at. % Co. Therefore, the amorphous FCZBN alloys should have the largest negative heat of mixing on replacement of Fe by 7.5 at. % Co, and



Figure 6. Plot of the apparent activation energy of glass transition of the amorphous FCZBN alloys as a function of Co content.

so have the densest random-packed structure. To confirm the above qualitative analysis, apparent activation energies of glass transition ( $E_g$ ) of the amorphous FCZBN alloys were measured by using Kissinger's method [23], and were plotted in figure 6 as a function of Co content. Figure 6 shows that the Co dependences of  $E_g$  are similar to those of  $T_{rg}$  in figure 2(c). It is known that  $E_g$  is related to the atomic diffusion ability of amorphous alloys near glass transition [24–26]. The higher the apparent activation energy, the more difficult the atomic diffusion, implying smaller atomic free volume and a denser atomic random-packing of the amorphous alloy [27, 28]. Therefore, the results of both figures 2(c) and 6 indicate that the amorphous FCZBN alloy with higher  $T_{rg}$  has a denser atomic random-packed structure, in agreement with experimental results reported previously [10–13, 29]. The density of the amorphous FCZBN alloy is much higher near a Co content of 7.5 at. % compared with other Co contents. As such, the interatomic distance of the Fe–Fe pair in the amorphous FCZBN alloys decreases with increasing  $T_{rg}$  and greatly decreases near 7.5 at. % Co.

Previous studies have demonstrated that the exchange interaction of Fe–Fe atomic pairs in Fe-rich Fe-based amorphous alloys decreases as the interatomic distance of an Fe–Fe couple reduces [30–32], resulting in a decrease of  $T_c$ . A similar phenomenon was also observed in the amorphous Fe<sub>91-x</sub>Zr<sub>5</sub> $B_x$ Nb<sub>4</sub> (FZBN) [33], an alloy system similar to the amorphous FCZBN used in this study. For the amorphous FZBN alloys, the  $T_c$  increases on increasing the B atoms (or decreasing Fe content). This is due to the increase in interatomic distance of Fe–Fe pair caused by decrease in Fe content and incorporation of B atoms into the framework composed of Fe, Zr and Nb [34–37], which results in increase of exchange interaction of Fe–Fe pair with increasing B content. Therefore, the interatomic distance of Fe–Fe pair has an important influence on the exchange interaction in the Fe–Zr–B–Nb alloy system. For the amorphous FCZBN alloys, as mentioned above, interatomic distance of the Fe–Fe pair decreases with increasing  $T_{rg}$ . Therefore, if only the exchange interaction of the Fe–Fe couples is taken into account, the  $T_c$  should decrease with increasing  $T_{rg}$  in the amorphous FCZBN alloys and reach a minimum in the amorphous alloy with Co content of 7.5 at. %.

On the basis of the above analysis, it is concluded that the Co-content dependence of the  $T_c$  is determined by the competition of two kinds of changing behaviours of  $T_c$ ; one is that the  $T_c$  increases with increasing Co content, which is attributed to the fact that the exchange

interactions of Fe-Co and Co-Co atomic couples are stronger than that of the Fe-Fe couple [1, 5]; and another is that the  $T_c$  decreases with increasing  $T_{rg}$  or GFA, which is due to the decrease in exchange interaction of the Fe-Fe couple with decreasing interatomic distance of the Fe–Fe couple. Amorphous FCZBN alloys with Co contents of 0, 2, 4, 10 and 15 at. % and with values the same or close to  $T_{rg}$  of 0.58, have close interatomic distance of Fe–Fe pairs. Consequently,  $T_c$  behaviour is determined by Co content and increases with increasing Co content. However, for the amorphous FCZBN alloys with Co content of 3 and 12.5 at. %,  $T_{rg}$ is less than 0.58, and so they should have a larger interatomic distance of the Fe–Fe couple compared with those for  $T_{rg}$  of 0.58 under the same Co content, resulting in them having a higher  $T_{\rm c}$  compared with those having  $T_{\rm rg}$  of 0.58 and the same Co content, as shown in figure 5(c). For the amorphous FCZBN alloys with 5 and 7.5 at. % Co,  $T_{rg}$  is greater than 0.58, and so the interatomic distance for a Fe-Fe couple should be much smaller than that for amorphous FCZBN alloys with  $T_{rg}$  of 0.58, resulting in a decreasing value of the exchange interaction induced by decrease in the interatomic distance that is much larger than the increase in the value of the exchange interaction induced by an increase in Co content. Therefore, the amorphous FCZBN alloys with 5 and 7.5 at. % Co have a much lower  $T_c$  compared with amorphous FCZB alloys irrespective of the Co content of the amorphous FCZBN alloys, as shown in figure 5(c). On the basis of the conclusions mentioned above, the experimental phenomenon that the rate of decrease of the  $T_{\rm c}$  with  $T_{\rm rg}$  is different for different Co-content ranges shown in figure 5(b) can also be explained. For Co content ranging from 0 to 3 and 7.5 to 12.5 at. %, T<sub>rg</sub> increases with decreasing Co content, and both decrease of Co content and increase in  $T_{rg}$  result in a decrease of  $T_c$ . For Co content ranging from 3 to 7.5 and 12.5 to 15 at. %  $T_{rg}$  increases with increasing Co content; however, increase in  $T_{rg}$  results in a decrease of  $T_c$ , but increase of Co content increases the  $T_{\rm c}$ . Therefore, the decreasing rate is larger for Co content ranging from 0 to 3 and 7.5 to 12.5 at. % compared with the Co content ranging from 3 to 7.5 and 12.5 to 15 at. %.

Similar results were also reported for other Fe-rich Fe-based amorphous alloys [16, 17]; for example the Ta content dependence of the  $T_c$  is opposite to that of  $T_g$  and GFA in amorphous Fe<sub>56</sub>Co<sub>7</sub>Ni<sub>7</sub>Zr<sub>10-x</sub>Ta<sub>x</sub>B<sub>20</sub> and Fe<sub>56</sub>Co<sub>7</sub>Ni<sub>7</sub>Zr<sub>10-x</sub>Nb<sub>x</sub>B<sub>20</sub> alloys [16]. The bulk Fe<sub>60</sub>Co<sub>8</sub>Zr<sub>10</sub>Mo<sub>5</sub>W<sub>2</sub>B<sub>15</sub> amorphous alloy has the highest  $T_g$  and GFA in the Fe-based amorphous alloy to the best of our knowledge, but its  $T_c$  is the lowest in the reported Fe–Co-based amorphous alloys [15].

#### 4. Conclusions

The  $T_c$  of the amorphous FCZBN alloys show a sinusoid-like changing behaviour with increasing Co content, revealing two maxima at 3 and 12.5 at. % and a minimum at 7.5 at. %. This behaviour is almost completely opposite to Co-content dependence of the  $T_g$ ,  $T_x$  and  $T_{rg}$  over the same composition range.  $T_c$  decreases with increasing  $T_g$  and  $T_{rg}$ , but increases with increasing Co content. The Co-content dependence of the  $T_c$  is determined by competition of effects of Co content and  $T_{rg}$  on the  $T_c$ . The effect of Co content can be attributed to the exchange interactions of Fe–Co and Co–Co couples, which are stronger than those of Fe–Fe, resulting in an increase in  $T_c$  with increasing Co content. The effect of  $T_{rg}$  is due to decrease in the exchange interaction of Fe–Fe couple with decreasing interatomic distance, caused by an increase in the density of the amorphous FCZBN alloy with increasing GFA, which results in a decrease of  $T_c$  with increasing  $T_{rg}$ .

#### References

- [1] He S, He H, Shen B, Zhang H, Zhang S and Guo H 1999 J. Appl. Phys. 86 6301
- [2] Shen B, Zhang B, Liang B, Gong H and Huo H 1995 J. Magn. Magn. Mater. 140/144 351

- [3] Shen B, Xu R, Zhao J and Zhao W 1991 Phys. Rev. B 43 11005
- [4] Shen B, Yang L, Zhang J, Wo F, Ning T-S and Zhan W 1991 J. Magn. Magn. Mater. 96 335
- [5] Müller M, Grahl H, Mattern N, Kühn U and Schnell B 1996 J. Magn. Magn. Mater. 160 284
- [6] Fujimori H, Morita H, Obi Y and Ohta S 1977 Amorphous Magnetism II, Proc. 2nd Int. Symp. on Amorphous Magnetism ed R A Levy and R Hasegawa (New York: Plenum) p 393
- [7] O'Handley R C, Hasegawa R, Ray R and Chou C P 1976 Appl. Phys. Lett. 29 330
- [8] Mizoguchi T, Yamauchi K and Miyajima H 1973 Amorphous Magnetism ed H O Hooper and A M DeGraff (New York: Plenum) p 325
- [9] Chen H S, Sherwood R C and Gyorgy E M 1977 IEEE Trans. Magn. 5 1538
- [10] Greer A L 1993 *Nature* **366** 303
- [11] Wang W-H and Bai H Y 1998 J. Appl. Phys. 84 5961
- [12] Inoue A, Negishi T, Kimura H M, Zhang T and Yavari A R 1998 Mater. Trans. JIM 39 318
- [13] Zhang T and Inoue A 1998 Trans. JIM 39 857
- [14] Inoue A, Zhang T and Takechi A 1997 Appl. Phys. Lett. 71 464
- [15] Inoue A 2000 Acta Mater. 48 279
- [16] Inoue A, Koshiba H, Zhang T and Makino A 1998 J. Appl. Phys. 83 1967
- [17] Koshiba H, Inoue A and Makino A 1999 J. Appl. Phys. 85 5136
- [18] Chen H S, Sherwood R C and Gyorgy E M 1997 IEEE Trans. Magn. MAG-13 1538
- [19] Lu Z P, Tan H, Li Y and Ng S C 2000 Scr. Mater. 42 667
- [20] Turnbull D 1969 Contemp. Phys. 10 473
- [21] Waniuk T A, Schroers J S and Johnson W L 2001 Appl. Phys. Lett. 78 1213
- [22] de Boer F R, Boom R, Manens W C, Miedema A R and Niessen A K 1988 Cohesion in Metals (Amsterdam:
- North-Holland)
- [23] Kissinger H E 1957 Anal. Chem. **29** 1702
- [24] Chen H S 1968 J. Chem. Phys. 48 2560
- [25] Matusita K and Sakka S 1980 J. Non-Cryst. Solids 38&39 741
- [26] Lopez-Alemany P L, Vazguez J, Villares P and Jimenze-Garay R 2000 J. Non-Cryst. Solids 274 249
- [27] Beuche F 1959 J. Chem. Phys. 30 748
- [28] Turnbull D and Cohen M H 1961 J. Chem. Phys. 34 120
- [29] Peker A and Johnson W L 1993 Appl. Phys. Lett. 63 2342
- [30] Hasegawa R 1981 J. Appl. Phys. 52 1847
- [31] Severin C S, Chen C W, Belova A J and Lin M C 1981 J. Appl. Phys. 52 1850
- [32] Chien C L and Unruh K M 1991 Phys. Rev. B 24 1556
- [33] Yao B, Hu X, Si L, Tan H and Li Y 2001 Proc. Symp. of 2000 Fall Meeting Materials Research Society, vol 644, p L7.3.1
- [34] Hong R and Ryan D H 1995 Phys. Rev. B 51 15885
- [35] Saito N, Hiroyoshi H, Fukamuchi K and Nakagawa Y 1986 J. Phys. F: Met. Phys. 16 911
- [36] Kaul S N, Siruguri V and Chandra G 1992 Phys. Rev. B 45 12343
- [37] Polk D E 1972 Acta Metall. 20 485